

Synthesis of a Graphene-Containing Composite by Deposition of Oxygen-Free Graphene on Nanocrystallite CeO₂

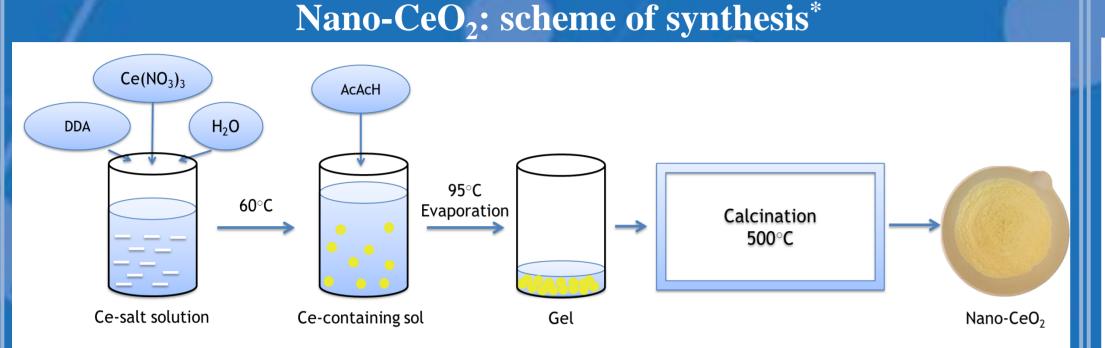
Ivan Ponomarev^{*}, Elena Trusova



Baikov Institute of metallurgy and material science of the Russian Academy of Sciences, IMET RAS, 49 Leninsky prospect, Moscow, 119334, Russian Federation *E-mail: IvanGforce@mail.ru

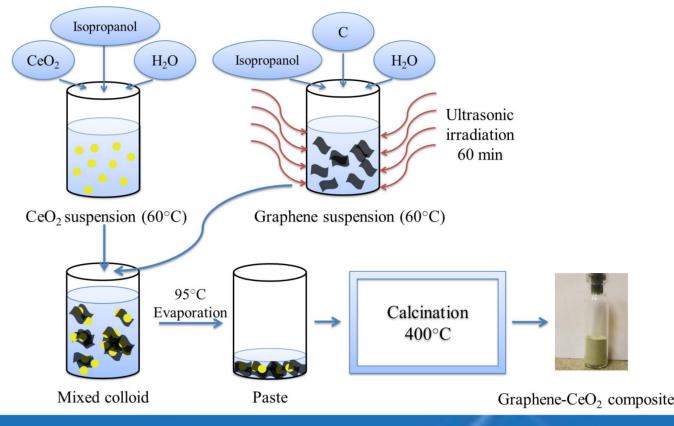
Currently, researchers and developers are interested in materials for supercapacitors, which are in demand for the development of small-sized electronic devices. Graphene containing materials based on CeO_2 are promising for these proposes. The content of graphene in the composite should not exceed 2 wt.% for the effective use of the material. Wherein the graphene sheets should contain a minimum amount of oxygen and be uniformly distributed in the material bulk. The fulfillment of these requirements is the most difficult problem to solve.

The aim of this work is to develop a method for the synthesis of hybrid structures based on nano- CeO_2 and oxygen-free graphene, which consists in depositing the latter onto metal oxide powder.



Nano-CeO₂ was synthesized by sol-gel method: 1) mixing of Ce(NO₃)₃ aqueous solution and dodecylamine (DDA) alcohol solution; 2) addition of acetylacetone (AcAcH) as complexing agent; 3) evaporation of the resulting sol to form a gel; 4) heat treatment at 500°C in oven for 1h.

Graphene-CeO₂ composite: scheme of synthesis*



Stages of synthesis:

1. obtaining an oxygen-free graphene suspension by ultrasonic exfoliation;

2. synthesis of CeO_2 nanopowder by sol-gel method;

3. mixing the aqua-alcohol suspension of synthesized nano- CeO_2 with the decanted graphene suspension;

4. evaporation of the mixed colloid to the paste state;

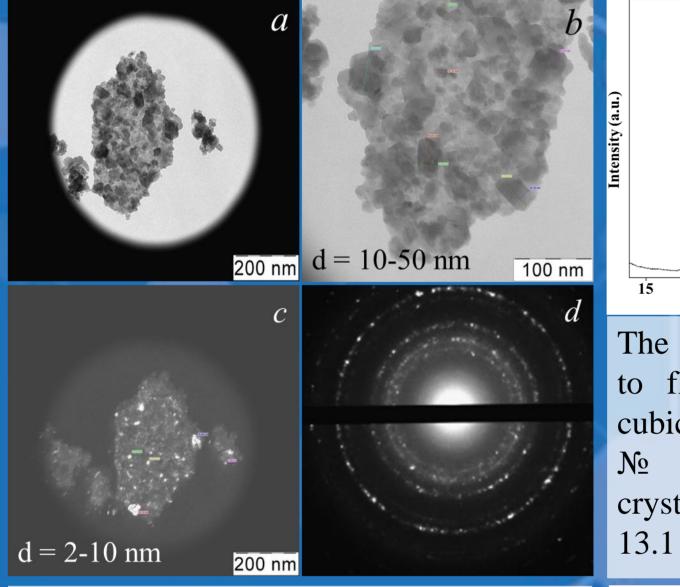
5. heat treatment at 400°C in oven for 1h.

*Ponomarev I V, Trusova E A, Afzal A M. Synthesis of graphene-CeO₂ nanocomposite using dodecylamine, Inorganic Materials: Applied Research, 2023, 14, 3, 868-875

*RU 2790846 C1 (2023). Trusova E A, Ponomarev I V, Afzal A M. A method of obtaining nanostructured composites based on oxygen-free graphene and aluminum or cerium oxides

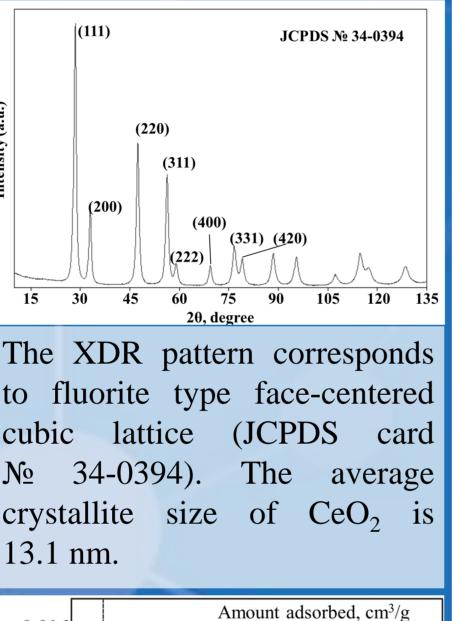
Nano-CeO₂: morphology and composition

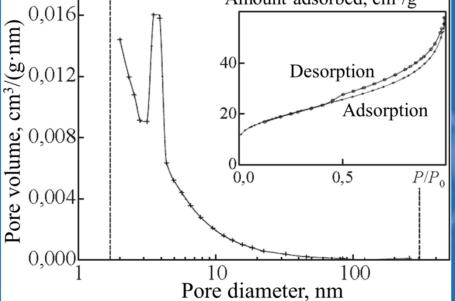
TEM data for the crystalline CeO_2 nanopowder: light-field (*a* and *b*) and dark-field (*c*) images and electron diffraction pattern for the sample area shown in Figs *a* and *c* (*d*).



According to N_2 adsorption-desorption data, the specific surface area of pure nano-CeO₂, calculated by the BET-method, was 68 m²/g. The average pore diameter was 4.5 nm which corresponds to a mesoporous system, while micropores provided less than 4% of volume and about 13% of the powder surface.

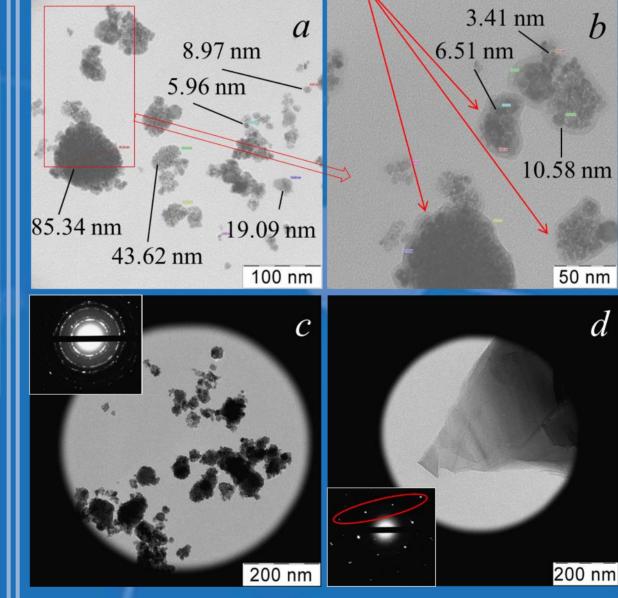
 CeO_2





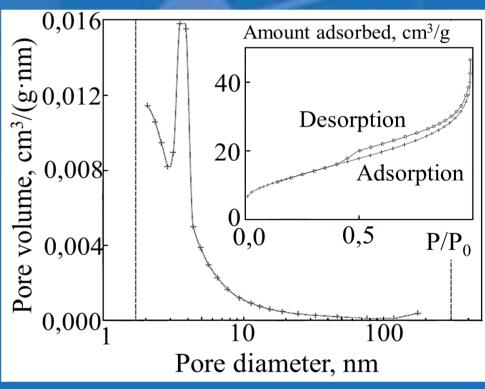
Morphology study of graphene-CeO₂ composite

Graphene shells – 4 nm



According to N_2 adsorptiondesorption data, the specific surface area of graphene-CeO₂ composite was 45 m²/g. The average pore diameter was 4.7 nm which corresponds to a mesoporous system, while micropores provided less than 3% of volume.

Graphene-CeO₂ composite powder consisted of agglomerates with sizes not exceeding 100 (20-60) nm, which had graphene shells 3-6 nm thick (a,b). CeO₂ crystallites inside the agglomerates had sizes 3-11 nm, were randomly and not tightly packed and wrapped with graphene sheets (c, inset). The electron diffraction pattern of peripheral part of agglomerate, where graphene sheets mainly appear (d, inset) corresponds to multilayer graphene, and the ratio of the characteristic reflection intensities indicates that the graphene sheets had variable layering: multilayer sections were adjacent to single-layer ones.



Mechanism of graphene-CeO₂ composites formation from two suspensions

Oxygen-free Graphene

Conclusion:

Thus, a method has been developed for the synthesis of hybrid structures based on nanocrystalline CeO₂ and oxygen-free graphene with a content of the latter of no more than 1 wt.%. The developed method makes it possible to obtain the thinnest sheets of oxygen-free graphene and uniformly distribute them among the CeO₂ nanocrystallites. The proposed method for the synthesis of composite nanostructured powders based on graphene and CeO₂ can be used as the basis for an economical and environmentally friendly technology for the nanopowders production, which are in demand in the development of materials for small-sized electronic devices.

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